

Naming ionic compounds worksheet

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|--|---|
| 1. TiSe_2 | titanium (IV) selenide |
| 2. Cu_2O | copper (I) oxide |
| 3. V_3P_5 | vanadium (V) phosphide |
| 4. $\text{Co}(\text{ClO}_3)_2$ | cobalt (II) chlorate |
| 5. $\text{Pb}(\text{SO}_4)_2$ | lead (II) sulfate |
| 6. CuCrO_4 | copper (II) chromate |
| 7. $(\text{NH}_4)_3\text{PO}_4$ | ammonium phosphate |
| 8. $\text{Cu}(\text{NO}_2)_2$ | copper (II) nitrite |
| 9. KCN | potassium cyanide |
| 10. Zn_3N_2 | zinc nitride |
| 11. NaMnO_4 | sodium permanganate |
| 12. $\text{Co}_3(\text{PO}_4)_2$ | cobalt (II) phosphate |
| 13. Li_2O | lithium oxide |
| 14. $\text{Fe}(\text{CH}_3\text{COO})_3$ | iron (III) acetate |
| 15. $\text{Sr}(\text{OH})_2$ | strontium hydroxide |
| 16. $\text{Cd}(\text{NO}_3)_2$ | cadmium nitrate |
| 17. CuO | copper (II) oxide |
| 18. PbS | lead (II) sulfide |
| 19. $\text{Mn}(\text{ClO}_2)_4$ | manganese (IV) chlorite |
| 20. NH_4HSO_3 | ammonium bisulfite or ammonium hydrogen sulfite |