

Lesson 5 Worksheet: Ionic Formula Writing

1. Aluminum bromide	AlBr_3
2. Platinum (II) sulphide	PtS
3. Strontium sulfite	SrSO_3
4. Scandium oxide	Sc_2O_3
5. Titanium (IV) nitrite	$\text{Ti}(\text{NO}_2)_4$
6. Ammonium sulphate	$(\text{NH}_4)_2\text{SO}_4$
7. Lithium selenide	Li_2Se
8. Lead (II) hydrogen sulphate	$\text{Pb}(\text{HSO}_4)_2$
9. Sodium acetate	NaCH_3COO
10. Cesium chloride	CsCl
11. Cadmium (II) hydroxide	$\text{Cd}(\text{OH})_2$
12. Zinc phosphate	$\text{Zn}_3(\text{PO}_4)_2$
13. Barium chloride	BaCl_2
14. Tin (II) permanganate	$\text{Sn}(\text{MnO}_4)_2$
15. Lithium hypochlorite	LiClO
16. Gold (III) sulphate	$\text{Au}_2(\text{SO}_4)_3$
17. Sodium nitrite	NaNO_2
18. Chromium (III) chloride	CrCl_3
19. Potassium carbonate	K_2CO_3
20. Iron (III) bisulphate	$\text{Fe}(\text{HSO}_4)_3$