

DATE:

NAME:

CLASS:

CHAPTER 4**Binary Ionic Compounds****BLM 2-12**

Goal • Demonstrate your ability to write the names and formulas of binary ionic compounds.

What to Do

1. Complete the following table.

Name of Compound	Formula
(a) beryllium fluoride	
(b) sodium nitride	
(c) calcium sulphide	
(d) aluminum chloride	
(e) lithium oxide	
(f) magnesium nitride	
(g) gallium sulphide	
(h) barium bromide	